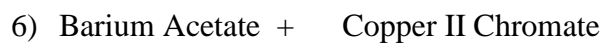
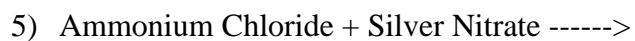
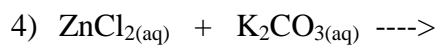
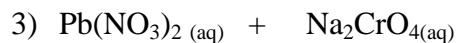
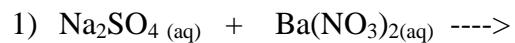


Predicting Single/Double Replacement Reactions (Chemical Reactions #4, #5); SC2a



Predict the products and balance the following single replacement reactions. If no reaction occurs write N.R. For transition metals use the “Activity of Metals Series” and the “solubility” tables that are in your composition books from the start of the unit.





7) Solutions of Silver Nitrate and ammonium chloride are mixed

8) Solutions of Lead II nitrate and Sodium Chloride are mixed

9) Solutions of zinc sulfate and magnesium chloride are mixed

10) Solutions of Ammonium phosphate and Zinc Chlorate are mixed

