Honors Chemistry: Chemical Reactions Podcasts 6 Problem Set- Net Ionic Equations

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For the	following	reactions	write the
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- a. Complete Equation (Indicate states): Balance
- b. Ionic Equation
- c. Net Ionic Equation
- 1. $ZnCl_2(aq) + Na_2S(aq) \rightarrow ZnS(s) + NaCl(aq)$
- 2. Ammonium phosphate combines with silver nitrate.

3. Magnesium Nitrate + Potassium phosphate → Magnesium Phosphate (s) + Potassium Nitrate

4. Lead(II) Acetate + Potassium Iodide

5. Sodium carbonate is mixed with hydrobromic acid.

6. Hydroiodic acid is mixed with sodium hydroxide.

7. Lithium sulfite is added to aqueous calcium nitrate.